RESTART (MAY 2022- APRIL 2025)

REcycling of Spent Li-ion baTteries and end-life photovoltAic panels: from the development of metal Recovery processes to the implementation of a sTart-up







Consortium

The partners were selected considering the complementarity of their expertise and the international multidisciplinary to achieve the objectives of the project

Project coordinator:

I. Saadoune, Cadi Ayyad University (UCA),
 Morocco

Project partners:

- Aalto University (Finland),
- Centre Européen de Recherche et d'Enseignement en Géosciences de l'Environnement - CEREGE (France),
- Mohammed VI Polytechnic Universitý -UM6P, and Green Energy Park (Morocco),
- Babeş-Bolyai University (Romania)
- King Salman International University (Egypt)

Aim of the project

The main objective of RESTART Project is to implement a full value chain for recycling End-of-Life (EoL) LiBs and PV, shifting from linear economy to circular economy, thus reducing waste disposal as well as minimizing dependence on critical minerals. The specific objectives are: **Collect**; **Recycle**; **Implement**; **Coordinate**.

The targeted metals: Co, Ni, Li, Mn are considered as strategic ones for the deployment of an effective energy transition

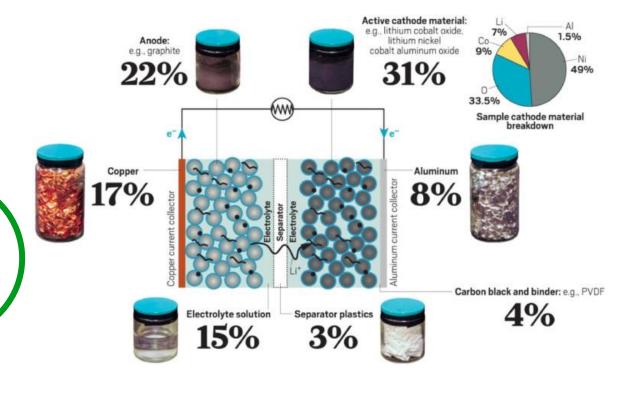
Relevance vs MARs

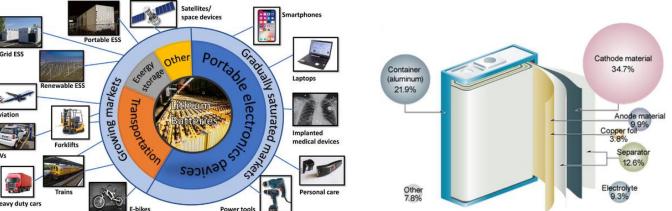
RESTART project's deliverables are in accordance with the following outcomes and impacts of MAR 2:

- Map of the EoL/OoS component value chain
- Proposal of methods for EoL/OoS component recycling
- Identification of second life components with a benefit for African countries
- Dissemination of acquired knowledge
- Creation of jobs
- Promotion of environmental and ecological sustainability

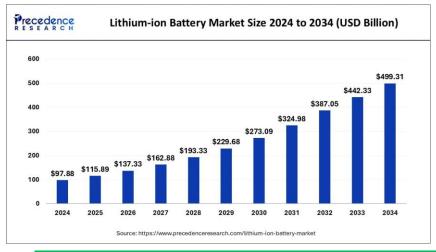








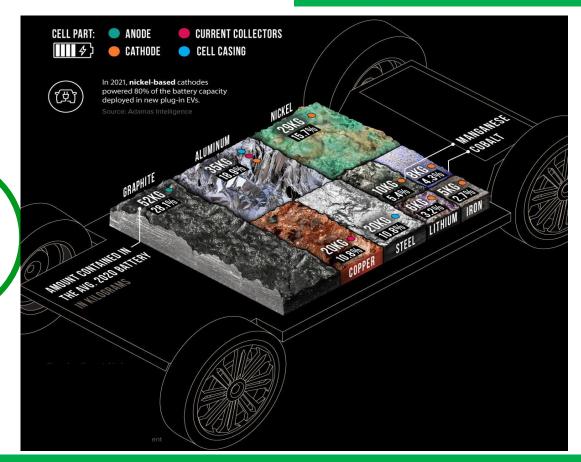


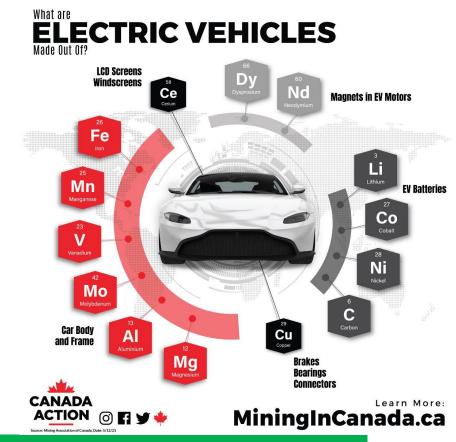


- 1. REDUCING THE ENVIRONMENTAL
 IMPACT (Groundwater; Refining
 Process (Mining))
- 2. ECONOMIC OPPORTUNITY

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Key Minerals in an EV Battery for instance

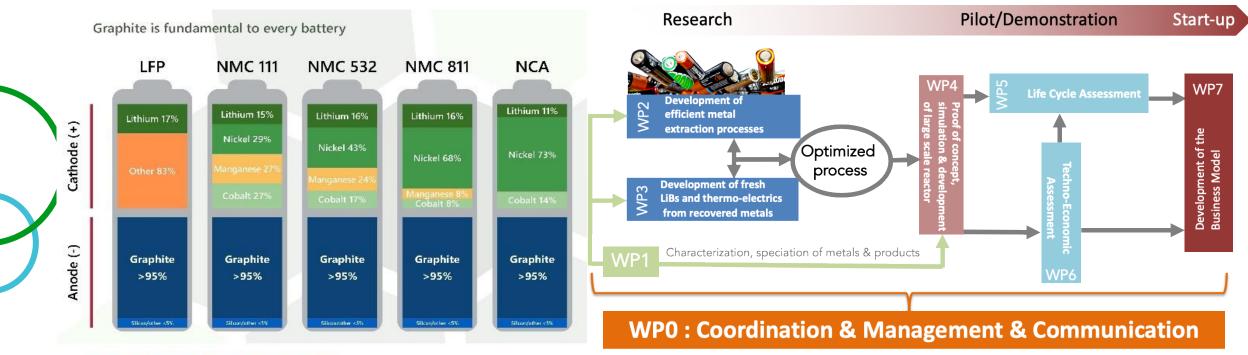




- Lithium-ion Batteries harness the properties of various minerals to power electric vehicles
- The cells in the average LiB Battery with a 60 KWh capacity contain around 185 Kg of Minerals (without considering the chemicals existing in binder, electrolyte, separator and pack casing



COMMERCIALIZED Li-ion BATTERIES



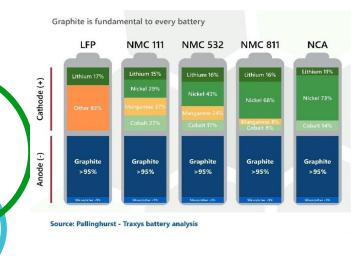
Source: Pallinghurst - Traxys battery analysis

RESTART Project is to implement a full value chain for recycling (EoL) LiBs and PV covering all aspects including: Technology; Sustainability and Economic

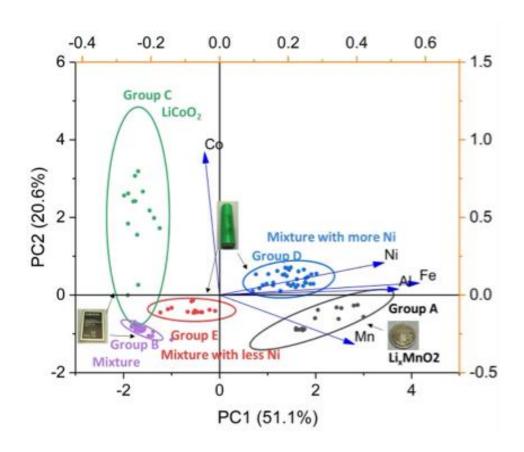


First Step: Determining the Composition and Chemical Forms of Elements in End-of-Life Batteries

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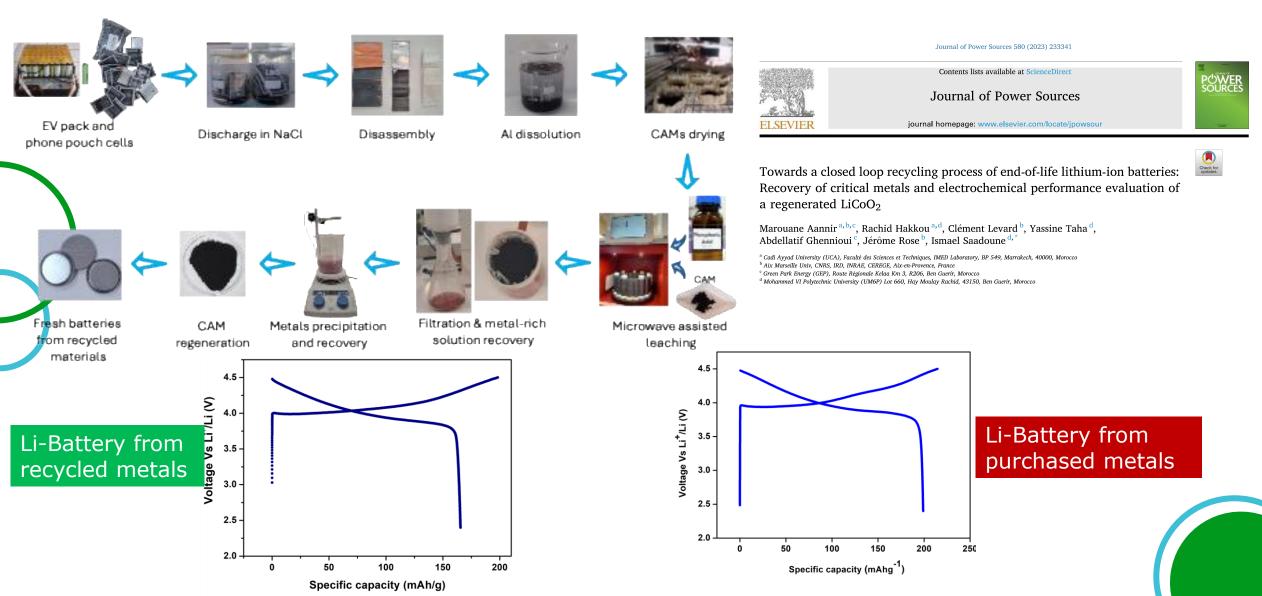


A Rapid Method to Identify the Cathode Type of End-of-Life Lithium-Ion Batteries by X-ray Fluorescence Spectroscopy for Optimizing Battery Recycling





Developed recycling processes for cathodes: Hydrometallurgy



Developed recycling processes for cathodes: Molten Salt

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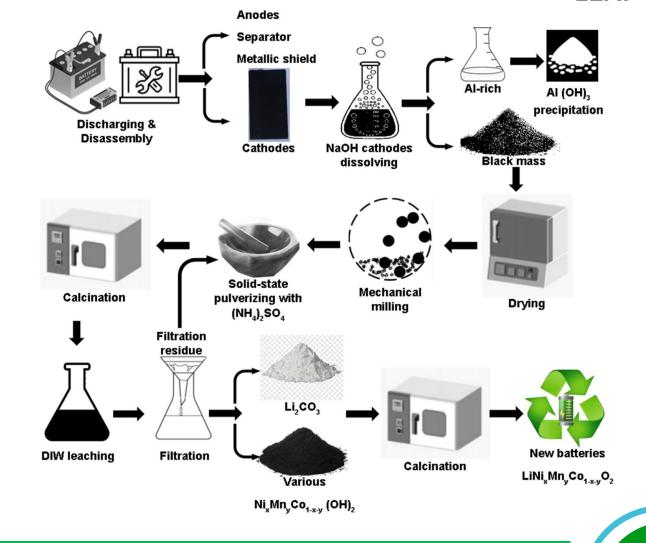


Comparative performance analysis of NMC cathodes elaborated from recovered and commercial raw materials: A low-temperature molten salt approach for extracting critical metals from end-of-life lithium-ion batteries

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- Green Park Energy (GEP), Route Régionale Kelaa Km 3, R206 Ben Guerir, Morocco Aix Marseille Univ, CNRS, IRD, INRAE, CEREGE, Aix-en-Provence, France
- Mohammed VI Polytechnic University (UM6P), GSMI, Lot 660, Hay Moulay Rachid, 43150 Ben Guerir, Morocco
- ^e Mohammed VI Polytechnic University (UM6P), ACER, Lot 660, Hay Moulay Rachid, 43150 Ben Guerir, Morocco





The leaching efficiencies reached 95.4%, 94.8%, 93.4%, and 92.6% for Li, Mn, Ni, and Co, respectively. The lithium carbonate obtained through ion-exchange exhibited a pure phase



Developed recycling processes for cathodes: Microwave using Phosphoric Acid

Patent draft

Abstract

A method is described for recycling lithium-ion batteries, involving disassembling the lithium-ion batteries into anodes, cathodes, separator, plastic case, and metallic case, and immersing the cathodic sheets in a sodium hydroxide aqueous solution. The filtrated battery residues are fed into a dryer. The dried residues are ground into fine particles, and mixed with an acid and reduction agent to form a slurry to leach metal oxides using microwave. The leachate is filtered to remove non-leachable metals and residue before being transferred to a precipitation tank. The leachate was brought to pH=6.3 to remove soluble aluminum contamination. Cobalt, nickel, and manganese were coprecipitated directly as hydroxide. Lithium was precipitated as lithium carbonate adding sodium carbonate to the remaining liquor and heating up. The precipitated hydroxide mixture of nickel, cobalt, and manganese and lithium carbonate were dried, ground together, and calcinated. The calcinated product was then used as cathodic active material.

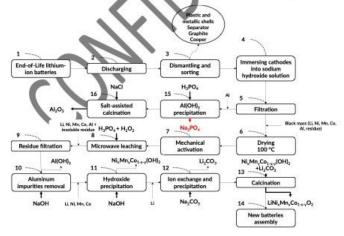
Technical field

[0001] It is provided a process for lithium-ion batteries.

Background

[0002] Recycling used lithium-ion batteries is an important aspect of managing the waste generated by these batteries, neutralizing the environmental impact of their production, and providing alternative sources of battery materials at affordable cost. Several recycling techniques were used, yet did not much the full requirements, either by presenting a significant environmental impact or not reaching the economic target by failing to recover many valuable metals.

[0003] Pyrometallurgical routes employ elevated temperatures to extract different elements from spent lithium-ion batteries, with anodic graphite and any organic components acting as the reducing agent and heat source. Heavier metals such as nickel, cobalt, and manganese are recovered as alloys and then further treated to do the separation. Lithium on the other hand is lost in the slag which decreases the overall economic of the processes.



Patent submitted in collaboration with IRESEN

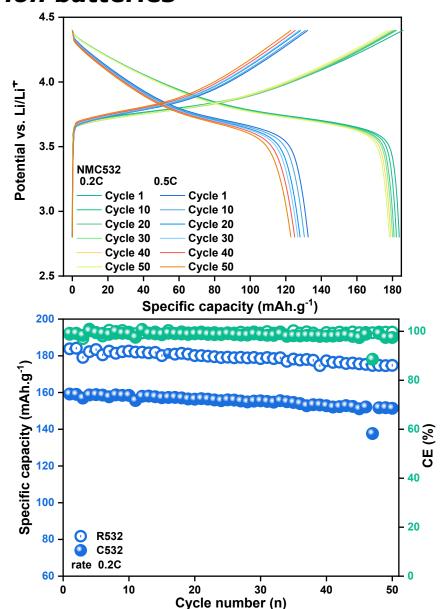
Regenerated products performance

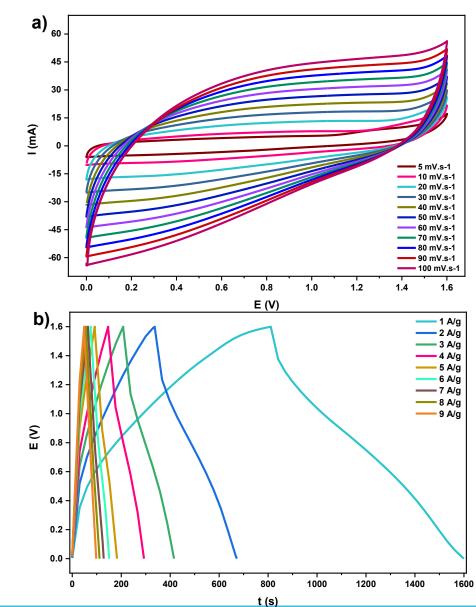
Cathodic materials:

• In Li-ion batteries





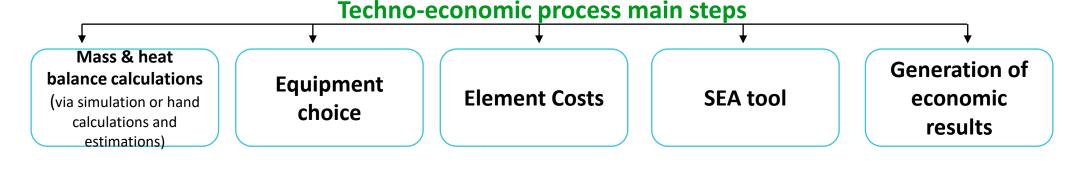




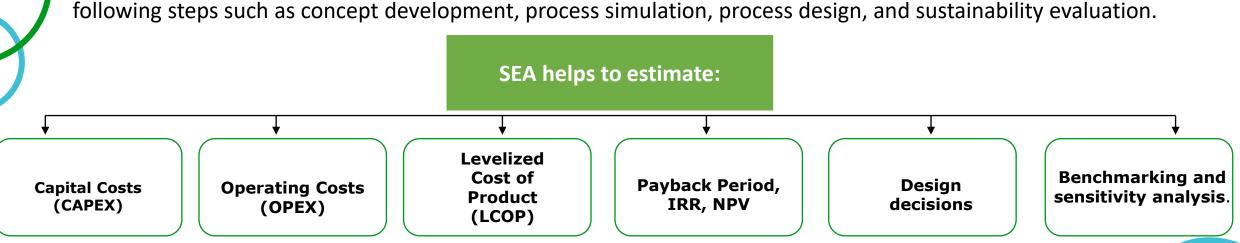
Techno-economic validation



Methodology: process design and SEA tool



Standardized Economic Assessment Tool (SEA) is an Excel-based framework designed to conduct technoeconomic assessments of chemical and energy processes. It integrates into the process development cycle, following steps such as concept development, process simulation, process design, and sustainability evaluation.

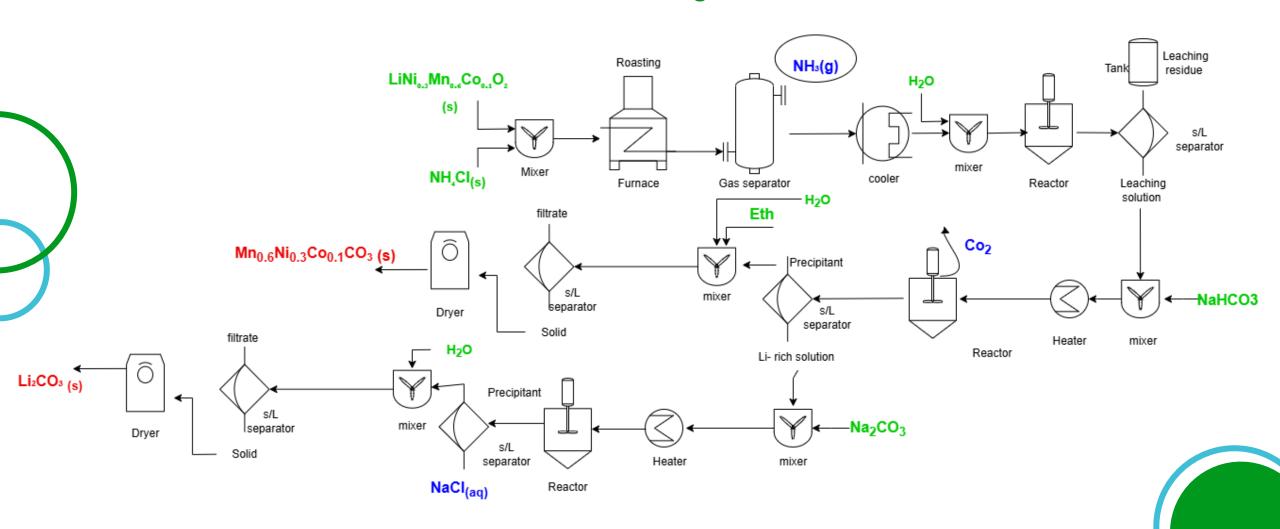


Techno-economic validation

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Molten salts: Applied on NMC

Process design



Techno-economic validation



Molten salts: Applied on NMC

Cash flow analysis

Economic Indicator	LCOP	
LCOP	3.822	\$/kg
NPV	3.3	M\$
IRR	8	%

LCOP: Levelized Cost of Product

NPV: Net Present Value **IRR**: Internal Rate of Return

Economic Analysis Assumptions		Economic Analysis Input/Output	
Capacity Factor 1st year	25%	Price of Product	6.78 \$/kg
Capacity Factor (remaining)	90%	Discount rate	8 %
Plant Lifetime	25 years	Discounted Cash Flow	3.269128109 M\$
Construction period	3 years		

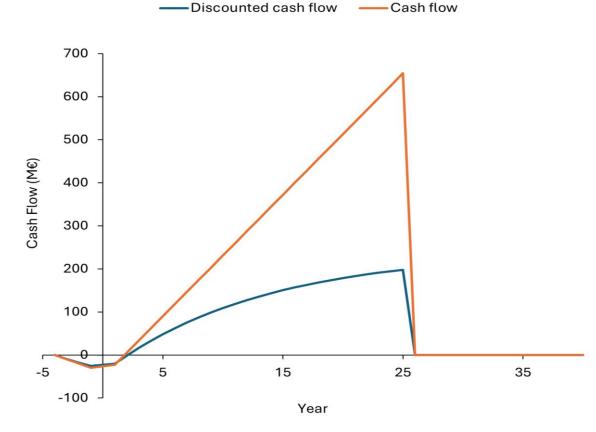
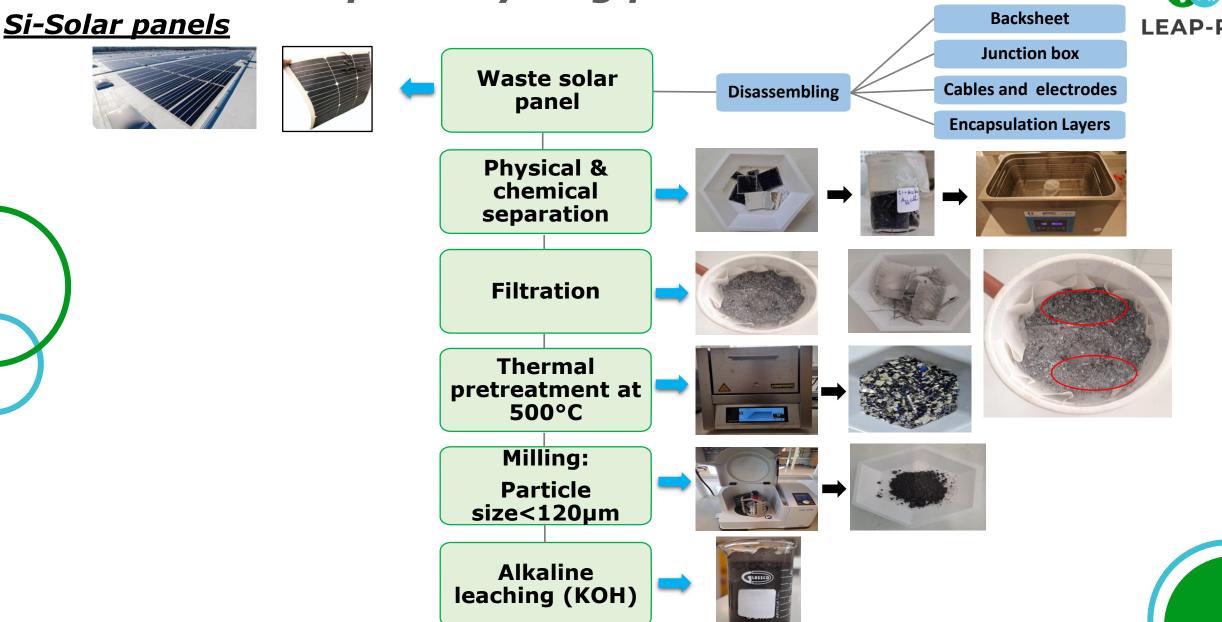


Figure 2. Cash Flow and Discounted Cash Flow Over 25-Year Project Lifetime

Developed recycling processes for anodes

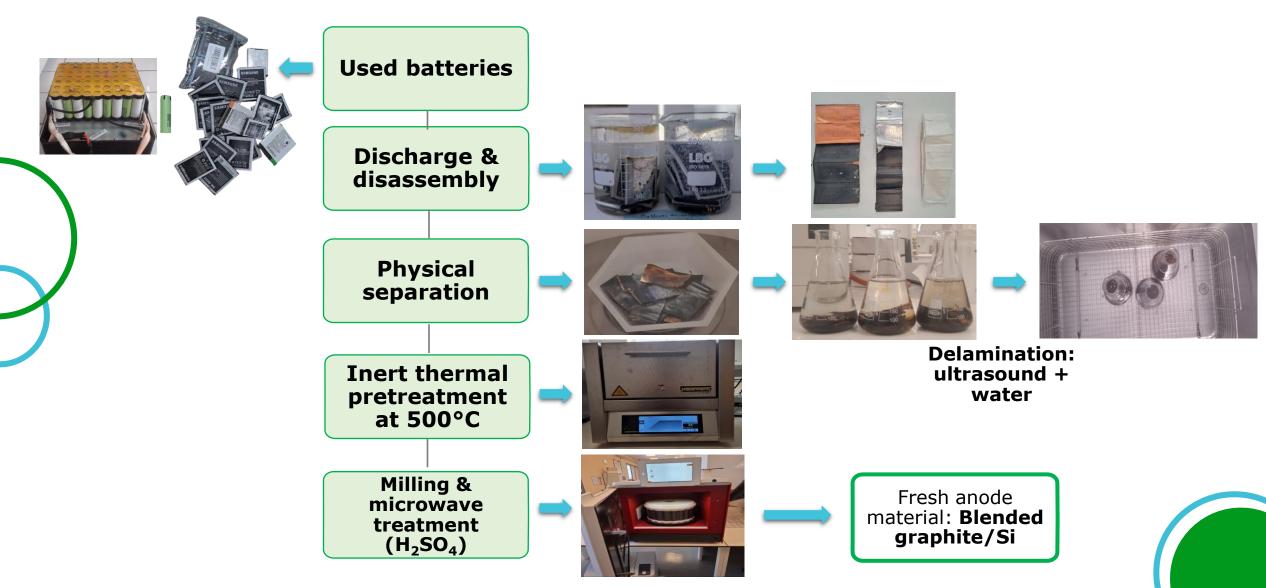




Developed recycling processes for cathodes

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Graphite recycling

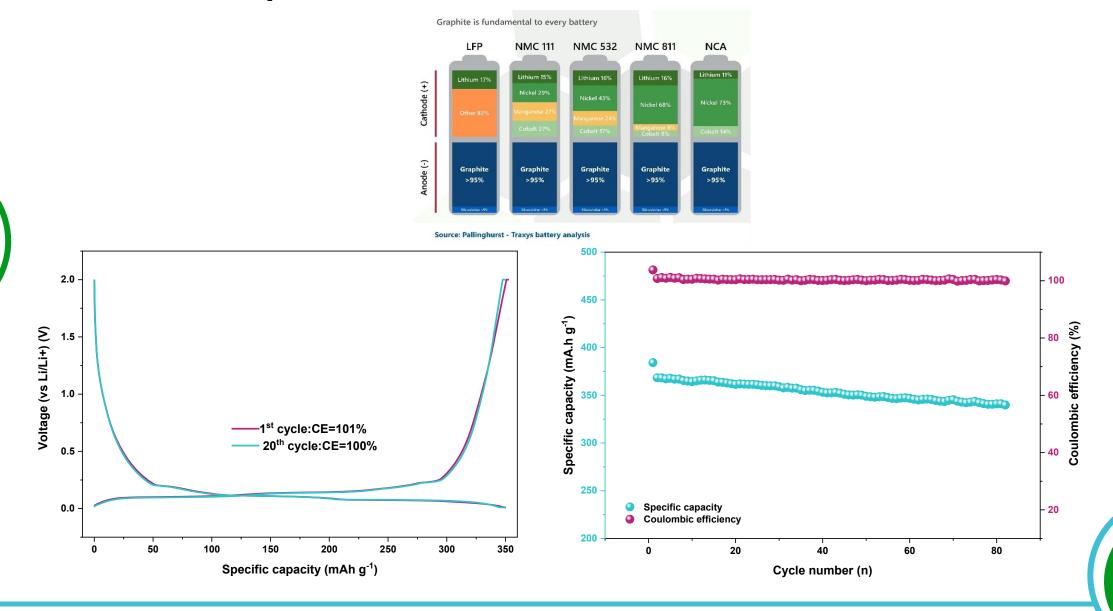


Regenerated products performance



Anodic materials: Graphite/Si

COMMERCIALIZED Li-ion BATTERIES



Achievements



2 PhD Thesis: Dr AANNIR (UCA-CEREGE); Dr EL MOUNAFIA (UM6P-GEP-UCA) (Job Creation)

5 Master Thesis

07 Scientific Publications (Q1, I.F. > 5)

02 Patents

Two funded Projects: 1. RCLIB (LEAP RE); 2. Morrocan Company

UNESCO Chair : e-waste Recycling (Polytech Montréal)



THANK YOU

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